

# WAVE PROPERTIES OF PARTICLES

1. De Broglie waves.
2. Waves of what?
3. Describing a wave.
4. Phase and group velocities.
5. Particle diffraction.
6. Particle in a box.
7. Uncertainty principle I.
8. Uncertainty principle II.
9. Applying the uncertainty principle.

# DESCRIBING A WAVE

## How fast do de Broglie waves travel?

This wave has the same velocity as the body it is associated with.

Let us call the de Broglie wave velocity  $v_p$  then:

$$v_p = v\lambda$$

$$E = h\nu = \gamma mc^2$$

$$\lambda = \frac{h}{\gamma m v}$$

$$v_p = v\lambda = \left(\frac{\gamma mc^2}{h}\right)\left(\frac{h}{\gamma m v}\right) = \frac{c^2}{v}$$

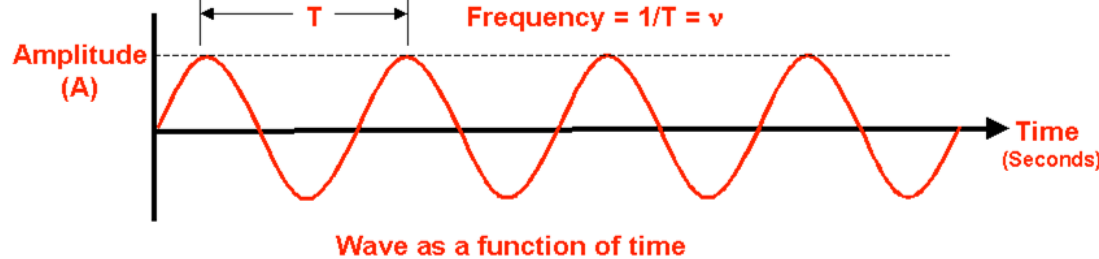
Faster  
than  
light!!



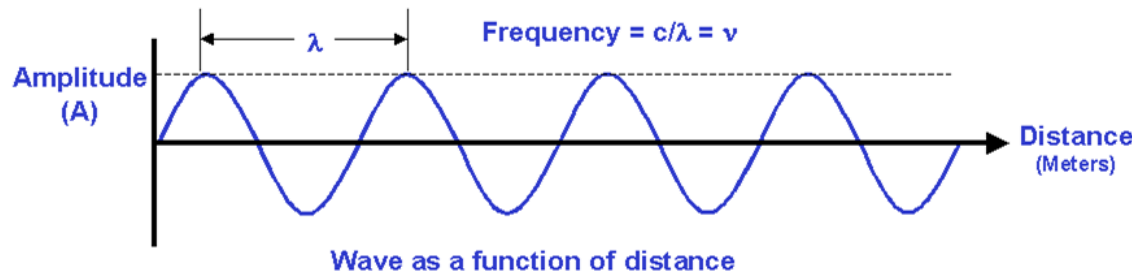
# DESCRIBING A WAVE

To resolve this we need to distinguish between **phase velocity** and **group velocity**.

Let us first look at the mathematical description of waves.



$$y = A \cos 2\pi vt$$



$$y = A \cos \frac{2\pi}{\lambda} x$$

$$y = A \cos 2\pi \left( vt - \frac{x}{\lambda} \right)$$

$$v_p = v\lambda$$

$$y = A \cos(\omega t - kx)$$

$\omega$  angular frequency

$k$  wave number =  $\frac{\omega}{v_p}$

# DESCRIBING A WAVE

## Remember...

The general formula of a wave is:

$$y = A \cos(\omega t - kx)$$

$\omega$  angular frequency

$$k \text{ wave number} = \frac{\omega}{v_p}$$

# PHASE AND GROUP VELOCITIES

## De Broglie waves...

→ the probability of finding a moving body at a particular place at a particular time.

## Can we represent the de Broglie wave by a simple wave equation?

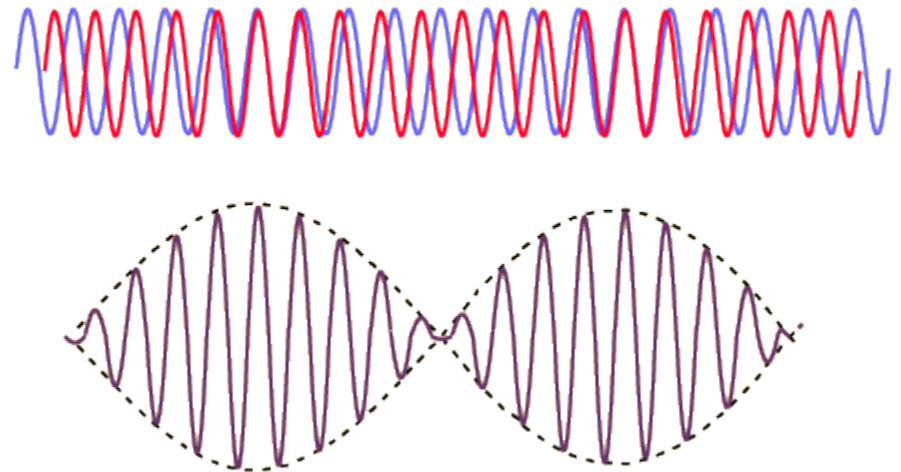
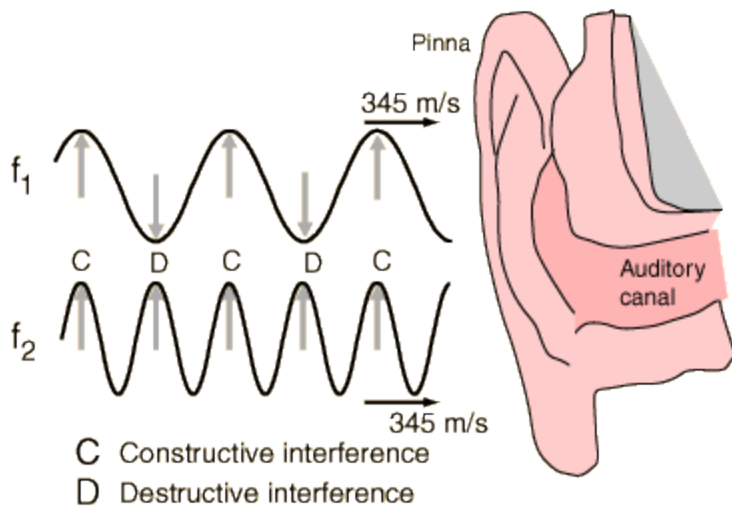
→ No! Simple wave formula of an indefinite series of waves all with the same amplitude  $A$ .  $y = A \cos(\omega t - kx)$

→ A wave representing a moving body must be a **wave packet** or **wave group**.

# PHASE AND GROUP VELOCITIES

Wave packet or wave group...

A familiar example is **beats**.

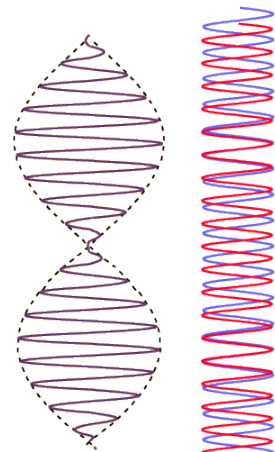


**How can we describe this mathematically?**  
**SUPERPOSITION!**

# PHASE AND GROUP VELOCITIES

## Superposition of what?

Individual waves of different wavelengths whose interference with one another results in a variation in amplitude that defines the group shape.



→ If the velocities of the waves are the same, the velocity of the wave group is the common phase velocity.

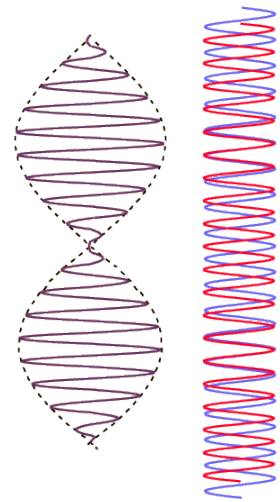
→ If the phase velocity varies with wavelength, the individual waves do not proceed together. ← **dispersion!**

→ the wave group has a velocity different from the phase velocities of the waves that make it up. ← de Broglie waves.

# PHASE AND GROUP VELOCITIES

How can we calculate  $v_g$ ?

Let us assume that the wave group arises from the combination of two waves that have the same amplitude  $A$  but differ by an amount  $\Delta\omega$  in angular frequency and  $\Delta k$  in wave number.



$$y_1 = A \cos(\omega t - kx)$$

$$y_2 = A \cos[(\omega + \Delta\omega)t - (k + \Delta k)x]$$

$$y = y_1 + y_2$$

$$\cos \alpha + \cos \beta = 2 \cos \frac{1}{2}(\alpha + \beta) \cos \frac{1}{2}(\alpha - \beta)$$

$$\cos(-\theta) = \cos(\theta)$$

$$y = y_1 + y_2 = 2A \cos \frac{1}{2}[(2\omega + \Delta\omega)t - (2k + \Delta k)x] \cos \frac{1}{2}(\Delta\omega t - \Delta kx)$$



# PHASE AND GROUP VELOCITIES

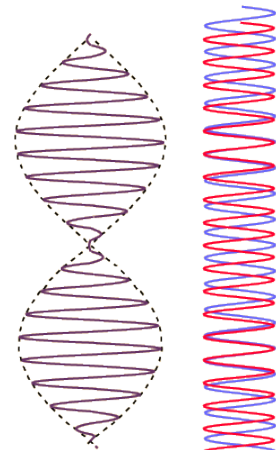
$$y = y_1 + y_2 = 2A \cos \frac{1}{2} [(2\omega + \Delta\omega)t - (2k + \Delta k)x] \cos \frac{1}{2} (\Delta\omega t - \Delta kx)$$

$\Delta\omega$  and  $\Delta k$  are small compared with  $\omega$  and  $k$ .

$$2\omega + \Delta\omega \approx 2\omega$$

$$2k + \Delta k \approx 2k$$

$$y = 2A \cos(\omega t - kx) \cos\left(\frac{\Delta\omega}{2} t - \frac{\Delta k}{2} x\right)$$



This is a wave of angular frequency  $\omega$  and wave number  $k$  that has superimposed upon it a modulation of angular frequency  $\Delta\omega/2$  and wave number  $\Delta k/2$ .

$$v_p = \frac{\omega}{k}$$

$$v_g = \frac{\Delta\omega}{\Delta k}$$

$$v_g = \frac{d\omega}{dk}$$

# PHASE AND GROUP VELOCITIES

For a de Broglie wave associated with a body of mass  $m$  moving with velocity  $v$ :

The angular frequency:

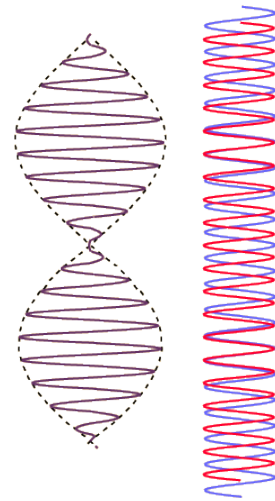
$$\omega = 2\pi\nu = \frac{2\pi\gamma mc^2}{h} = \frac{2\pi\gamma mc^2}{h\sqrt{1-v^2/c^2}}$$

The wave number:

$$k = \frac{2\pi}{\lambda} = \frac{2\pi\gamma m v}{h} = \frac{2\pi m v}{h\sqrt{1-v^2/c^2}}$$

Both  $\omega$  and  $k$  are functions of the body's velocity  $v$ .

**What is the group velocity of de Broglie waves?**



# PHASE AND GROUP VELOCITIES

What is the group velocity of de Broglie waves?

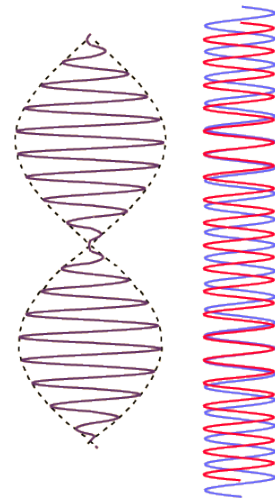
For a de Broglie wave associated with a body of mass

$$v_g = \frac{d\omega}{dk} = \frac{d\omega/dv}{dk/dv}$$

$$\frac{d\omega}{dv} = \frac{2\pi m v}{h(1-v^2/c^2)^{3/2}}$$

$$\frac{dk}{dv} = \frac{2\pi m}{h(1-v^2/c^2)^{3/2}}$$

$$v_g = v$$



*The de Broglie wave group associated with a moving body travels with the same velocity as the body.*

# PHASE AND GROUP VELOCITIES

**What is the phase velocity of de Broglie waves?**

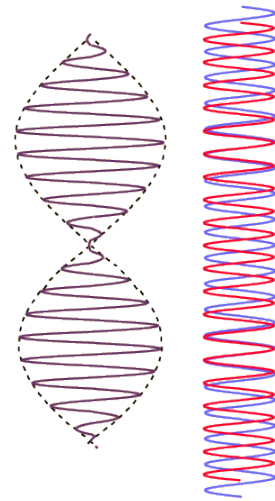
$$v_p = \frac{\omega}{k} = \frac{c^2}{v}$$

It exceeds both the velocity of the body and the velocity of light.

**How can this be justified?**

$v_p$  has no physical significance because the motion of the group not the individual correspond to the motion of the body and  $v_g < c$ .

→ De Broglie wave does not violate special relativity!



# PHASE AND GROUP VELOCITIES

## **Example 3.3:**

An electron has a de Broglie wavelength of  $2.00 \text{ pm} = 2.00 \times 10^{-12} \text{ m}$ . Find its kinetic energy and the phase and group velocities of its de Broglie waves.